



0040-4039(94)00846-9

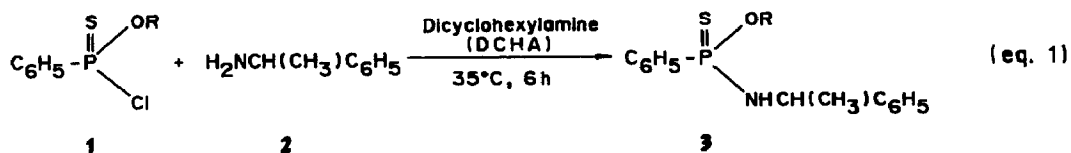
**A Simple Route to Chiral Phosphonothionates from  
 Diastereomeric Phosphoramidothionates.**

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**Abstract:** Optical isomers of phosphonothionates have been prepared by acid catalyzed alcoholysis of resolved phosphoramidothionates in high optical purity.

Chiral organophosphorus compounds, exhibit differential toxicity<sup>1</sup> and metabolism<sup>2</sup> in living organisms. Moreover, these compounds find application in elucidating organic and enzymatic reaction mechanisms<sup>3,4</sup>. The only procedure available to-date for the preparation of these compounds is through conversion of resolved phosphonothioic acids into their chlorides followed by reaction with phenols<sup>5</sup>. However, this method involves multisteps and consequently affords poor yields. Recently stereoselective P-O and P-N bond cleavage was adopted for preparation of chiral esters<sup>6</sup>. To our knowledge this methodology has never been considered for the synthesis of insecticides. In this communication the utility of stereospecific P-N bond cleavage for the synthesis of pure enantiomers of cyanofenphos (O-ethyl, O-4-cyanophenyl phenyl phosphonothionate) an insecticide and related compounds are described.

Reported procedure<sup>7</sup> provided good yield of intermediates 1a-c. This prompted us to examine the reaction with optically active isomers R(+) and S(-)-  $\alpha$ -methyl benzylamine 2 (eq. 1).

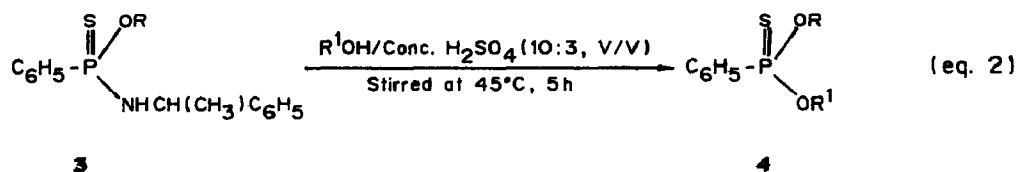


a : R = 4-CN-C<sub>6</sub>H<sub>4</sub>, b : R = 4-Cl-C<sub>6</sub>H<sub>4</sub>, c : R = C<sub>6</sub>H<sub>5</sub>

The following procedure for preparation of O-4-cyanophenyl phenyl phosphonamidothionate **3** is representative. To **1a** (6.0g, 20 mmol) taken in a flask containing dry benzene (50ml) was added slowly a solution of S(-)-isomer of **2** (2.4g, 20mmol) and DCHA (3.6g, 20mmol) in dry benzene (50 ml), with stirring. The reaction mixture was stirred at 35°C for another 6 h and DCHA salt thus formed was filtered. Solvent was removed under vacuum and the residual liquid so obtained was cleaned up by column chromatography on silica gel with eluent; benzene/hexane (8:2) Evaporation of collected fractions afforded **3** (6.3 g, 83% yield) as viscous liquid. This was dissolved in hexane/ether (10:3) and kept aside overnight which upon rubbing the sides of flask gave white crystalline solid **3a** (1.8 g, 28% yield).

Determination of specific rotation and  $^{31}\text{P}$  NMR by Perkin Elmer polarimeter model 241 and 90 MHz spectrometer model Jeol EX90 respectively showed hundred percent purity of diastereomer P(-)C(-). Isomer P(+ )C(+ ) was obtained by reacting R(+)-  $\alpha$ -methylbenzylamine with **1a** as described above. Though the synthesis of diastereomer of **3b, c** was done in a similar manner the resolution was obtained through fractional crystallization in petroleum ether (40-60°C). Microanalysis, IR,  $^1\text{H}$  NMR and  $^{31}\text{P}$  NMR spectroscopy was used to characterize the products<sup>8</sup>. Physical data and specific rotation of resolved isomers **3a-c** are summarized in table 1.

Acid catalyzed methanolysis/ethanolysis of **3a-c** resulted in products<sup>9</sup> **4a-e** through stereospecific P-N bond cleavage as represented (eq .2).



An evidence for high stereospecificity was brought by comparing observed specific rotation with reported values<sup>3</sup>. Products **4a-e** were characterized by IR and  $^1\text{H}$  NMR spectral data<sup>10</sup>. Physical data and specific rotation of enantiomers **4a-e** are given in table 1.

The present method thus provided an easy access to chiral isomers of cyanofenphos, an important neurotoxic esterase inhibitor.

Table 1. N-( $\alpha$ -methylbenzyl)phosphonamidothionate

Product No.	R	yield (%)	Diastereomer	$[\alpha]_D^{20}$ (C=5, CHCl <sub>3</sub> ) °C	$n_D$ °C	Molecular Formula
3a	4-CN-C <sub>6</sub> H <sub>4</sub>	26	P(+)-C(+)	+36.45	89-90	C <sub>21</sub> H <sub>19</sub> N <sub>2</sub> O <sub>2</sub> PS (378.43)
		28	P(-)-C(-)	-36.50	89-90	
3b	4-Cl-C <sub>6</sub> H <sub>4</sub>	27	P(+)-C(+)	+46.80	87-88	C <sub>20</sub> H <sub>19</sub> N <sub>2</sub> O <sub>2</sub> PSCl (387.85)
		24	P(-)-C(-)	-47.00	87-88	
3c	C <sub>6</sub> H <sub>5</sub>	25	P(+)-C(+)	+58.40	84-85	C <sub>20</sub> H <sub>20</sub> N <sub>2</sub> O <sub>2</sub> PS (353.42)
		27	P(-)-C(-)	-58.60	84-85	

Table 2. Optically Active Phosphonothionates

Product No.	R	R <sup>1</sup>	Yield (%)	$[\alpha]_D^{20}$ (C=4, CHCl <sub>3</sub> )	$n_D$ °C	Molecular Formula
4a	4-CN-C <sub>6</sub> H <sub>4</sub>	C <sub>2</sub> H <sub>5</sub>	63	(R):+37.7 [33.5, C=4.0] <sup>3</sup>	100-101	C <sub>15</sub> H <sub>14</sub> N <sub>2</sub> O <sub>2</sub> PS (303.31)
			66	(S):-38.1 (-33.7, C=3.0) <sup>3</sup>		
4b	4-CN-C <sub>6</sub> H <sub>4</sub>	CH <sub>3</sub>	60	(R):+46.7	60-61	C <sub>14</sub> H <sub>12</sub> N <sub>2</sub> O <sub>2</sub> PS (289.29)
			58	(S):-45.8		
4c	4-Cl-C <sub>6</sub> H <sub>4</sub>	C <sub>2</sub> H <sub>5</sub>	57	(R):+29.3	Liquid	C <sub>14</sub> H <sub>14</sub> O <sub>2</sub> PSCl (312.75)
			60	(S):-29.4		
4d	4-Cl-C <sub>6</sub> H <sub>4</sub>	CH <sub>3</sub>	64	(R):+37.8	Liquid	C <sub>13</sub> H <sub>13</sub> O <sub>2</sub> PSCl (299.73)
			65	(S):-38.1		
4e	C <sub>6</sub> H <sub>5</sub>	CH <sub>3</sub>	54	(R):+27.1	Liquid	C <sub>13</sub> H <sub>13</sub> O <sub>2</sub> PS (264.28)
			56	(S):-27.7		

3 Specific rotations under parentheses are reported values.

**Acknowledgement.** We are thankful to Dr R.V. Swamy, Director, DRDE, Gwalior for his valuable advise through out this work. We also thank to Mr. Meehir Palit, Mr Lakhi Ram and Km Mamta Pandey for providing the analytical and spectral data and Mr.G.R. Khanwilkar for secretarial assistance.

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8. **3a**: IR (KBr) 3360, 2240, 1225  $\text{cm}^{-1}$ ;  $^1\text{H}$  NMR (90MHz,  $\text{CDCl}_3$ ):  $\delta$  1.46 (d, 3H, J=6.49 Hz), 3.7 (t, 1H, J=9.74 Hz), 4.58 (m, 1H, NH), 7.07-8.01 (br, m, 14H);  $^{31}\text{P}$  NMR (90MHz,  $\text{CDCl}_3$ ):  $\delta$  75.2; Anal. Calcd: C, 66.65; H, 5.06; N, 7.40 Found: C, 67.04; H, 5.08; N, 7.13  
**3b**: IR (KBr) 3300, 1200  $\text{cm}^{-1}$ ;  $^1\text{H}$  NMR (90MHz,  $\text{CDCl}_3$ ):  $\delta$  1.47 (d, 3H, J=6.83 Hz), 3.53 (t, 1H, J=9.40 Hz), 4.51 (m, 1H, NH), 7.05-8.01 (br, m, 14H);  $^{31}\text{P}$  NMR (90MHz,  $\text{CDCl}_3$ ):  $\delta$  76.61; Anal. Calcd: C, 61.93; H, 4.94; N, 3.61 Found: C, 61.90; H, 4.95; N, 3.28  
**3c**: IR (KBr) 3300, 1200  $\text{cm}^{-1}$ ;  $^1\text{H}$  NMR (90MHz,  $\text{CDCl}_3$ ):  $\delta$  1.43 (d, 3H, J=6.89 Hz), 3.53 (t, 1H, J=9.79 Hz), 4.51 (m, 1H, NH); 7.00-8.01 (br, m, 15H);  $^{31}\text{P}$  NMR (90MHz,  $\text{CDCl}_3$ ):  $\delta$  73.98; Anal. Calcd: C, 67.97; H, 5.70; N, 3.96. Found: C, 67.70; H, 5.55; N, 3.47
9. **4a**: P(-)C(-) diastereomer of **3a** (0.4g, 1.05 mmol) was stirred with a mixture (20ml) of ethanol/Conc.  $\text{H}_2\text{SO}_4$  (10:3 v/v) at 45°C for 5h. Reaction mixture was diluted with water (20ml) and extracted with two portions of 30ml each with ethyl ether. Ether layer was first washed with 3%  $\text{NaHCO}_3$  solution (10ml) and twice with water (each 10ml). Ether was evaporated off and the crude product was purified by column chromatography on silica gel using benzene/hexane (8:2). Evaporation of solvent under reduced pressure yielded **4a** (0.21 g, 66% yield) m.p. 100-101°C.
10. **4a**: IR (KBr) 2230, 1215, 1040  $\text{cm}^{-1}$ ;  $^1\text{H}$  NMR (90MHz,  $\text{CDCl}_3$ ):  $\delta$  1.37 (t, 3H, J=7.1 Hz), 4.28 (m, 2H), 7.14-8.05 (br, m, 9H)  
**4b**: IR (KBr) 2230, 1220, 1050  $\text{cm}^{-1}$ ;  $^1\text{H}$  NMR (90MHz,  $\text{CDCl}_3$ ):  $\delta$  3.85 (d, 3H, J=14.18 Hz); 7.13-8.01 (br, m, 9H)  
**4c**: IR (KBr) 1210, 1035  $\text{cm}^{-1}$ ;  $^1\text{H}$  NMR (90MHz,  $\text{CDCl}_3$ ):  $\delta$  1.42 (t, 3H, J=7.05 Hz), 4.24 (m, 2H), 7.00-8.12 (br, m, 9H)  
**4d**: IR (KBr) 1210, 1040  $\text{cm}^{-1}$ ;  $^1\text{H}$  NMR (90MHz,  $\text{CDCl}_3$ ):  $\delta$  3.82 (d, 3H, J=13.85 Hz), 7.03-8.11 (br, m, 9H)  
**4e**: IR (KBr) 1200, 1040  $\text{cm}^{-1}$ ;  $^1\text{H}$  NMR (90MHz,  $\text{CDCl}_3$ ):  $\delta$  3.82 (d, 3H, J=12.92 Hz), 7.01-8.11 (br, m, 10H)

(Received in UK 15 April 1994; accepted 29 April 1994)